MAE 119 Professor G.R. Tynan Winter 2013 Quiz 3 Open Book/Open Notes

In our simple carbon balance model, we assumed that the flux of C between the atmosphere and land & ocean was proportional to the deviation of carbon concentration from the equilibrium value. The atmospheric carbon balance model could then be written as

$$\frac{\partial}{\partial t} \delta M_C(t) = Q_C(t) - \frac{\delta M_C}{t_{net}}.$$

Where $t_{net} = \frac{t_1 t_2}{t_1 + t_2}$ denotes the effective timescale for C exchange with the

Earth surface and oceans, $\delta M_c(t)$ denotes the deviation of the atmospheric carbon content away from the equilibrium value, and $Q_c(t)$ is the carbon injection rate from fossil fuel combustion.

Suppose the carbon source, $Q_C(t<0)=8$ GigaTonnes/year, is constant for time t<0, and then at t=0 the carbon source injection rate is doubled so that for t>0 $Q_C(t>0)=16$ GigaTonnes/year. The effective absorption timescale is given as $t_{net} = 100$ years.

- a) For very early times (i.e. t << 0) what is δM_c ? 5 POINTS
- b) For very late times (i.e. t >> 0) what is δM_c ? 5 POINTS
- c) Sketch the time evolution of δM_c . How long will it take for δM_c to get within about 70% of its final value after the change in injection rate at t=0? One significant figure will suffice. 5 POINTS
- d) For very late times (i.e. t >> 0) what will be the value of the IR transmission coefficient *relative* to the value it had at very early times (i.e. t<<0)? 5 POINTS</p>

EXTRA CREDIT (10 POINTS):

Find the time evolution of $\delta M_C(t)$. *Hint:* You don't have to solve the model ODE. Instead, note that this model is *linear* and so solutions and C sources can be superimposed.

a) $\delta M_c = Q_c b_{off}$ for thescase. Thus $\delta M_c = 8.100 = 800$ GTonnes b) $\delta M_c = Q_c b_{off}$ " $M_c = M_c b_{tro}$ $\delta M_c = 1600$ GTonnes $\delta M_c = Q_c b_{off}$ $\delta M_c = 1600$ GTonnes $\delta M_c = M_c b_{off}$ - 800 GTonnes It will take about 1 x tags = 100 years for - 800 GTonnes It will take about 1 x tags = 100 years for the SMc value to get to 270% y its final value C)

Ţ 162 87 CON 6 t=0 Q=unst =8 Q gian -0 Solutin: Solut SM2= Red weet tret) 12= Octnot (1-e 2 >0 SM2=0 tco total SM = Qctnot + Qctnet (1-e^t/tnet) t>o Restruct (2 - e-t/treet 8Me = SMC thet', tco Coop Ail Social

Quiz 3 (cont'd) (d) $\beta = exp(-\sigma n_{gg} d)$ Ngg = Mgg where Mgg is molecular Mgg V mass of CO2 and V is atmospharic Volum $H_{us} = \frac{e_{xp}\left(-\left(\frac{\sigma d}{m_{gg}}\right)M_{gg}\right)}{e_{xp}\left(-\left(\frac{\sigma d}{m_{gg}}\right)M_{gg}\right)}$ Birltro exp(-(50)Mggltro) $= e_{xp} \left(- \frac{\sigma d}{m_{gg}} V \left(M_{gg} \right) - M_{gg} \left(t_{xo} \right) \right)$ but Mgg = Z Mgg / teco $\frac{\delta}{\delta} \frac{\beta_{1R}}{\beta_{1R}} = \exp\left(-\left(\frac{\delta d}{m_{N}}\right) M_{99}\right) + \left(\frac{\delta}{\delta}\right) = \beta_{1R} + \left(\frac{\delta d}{\delta}\right) + \left(\frac{\delta d}$

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